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the mechanism of sulfur oxidation in the soil, both in the absence and in the presence of small and large amounts of rock phosphate. The sulfur and phosphate were added to the soil and well mixed.

### **THE CHEMISTRY OF THE OXIDATION OF SULFUR BY**

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sulfur compounds, phosphites and phosphines, which reduce hydroperoxides into the more stable alcohols. Whether or not these compounds can be used in foods has not

### **LIPID OXIDATION - USDA**

2. Dioxirane oxidation of the sulfur atom of sulfur-containing organic compounds 2.1. Dioxirane oxidation of the thianthrene 5-oxide 2.2. Oxidation of thioethers and thiol esters 2.3. Chemoselective oxidation of the sulfur atom 2.4. Stereoselective sulfoxidation of sulfides with dimethyldioxirane 2.5.

### **Dioxirane oxidation of sulfur-containing organic compounds**

Sulfur compounds and high concentrations of carbon black inhibit oxidation of hydrocarbon polymers primarily by this mechanism. Other types of peroxides, e.g., diperoxides and cyclic peroxides, which may be formed during autoxidation of certain hydrocarbon polymers, are much more stable to decomposition by sulfur compounds.

### **Mechanisms contributing to the autoxidation of hydrocarbon**

We now have studied the general reactions of organic sulfur compounds bearing S-S linkage, thiols and sulfinates with superoxide anion. This paper describes general features of the reactions of various organic sulfur compounds with O<sub>2</sub><sup>-</sup>.

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