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To the left of the symbol we find the atomic mass (A) at the upper corner, and the atomic number (Z) at the lower corner. Examples: ${}^1_1\text{H}$ ${}^{12}_6\text{C}$ ${}^{23}_{11}\text{Na}$ A Symbol Z Electron trade constitutes the currency of chemical reactions. The number of electrons in a neutral atom (that is, the atomic number) gives the element its unique identity.

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Tested Thomson's theory of atomic structure with the gold foil experiment in 1910.

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