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This note is intended primarily as an introduction to inorganic chemistry for undergraduate students. Topics covered will include: atomic structures, periodic trends, symmetry and group theory, inorganic solids, molecular orbital theory, molecular structure, acid-base chemistry, and band theory.

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and inorganic molecules. 3 Objective 1 Inorganic molecules: Relatively small, simple molecules that usually lack C (a few have one C atom). Examples: CO₂, NH₃, H₂O, O₂, H₂ 4 Objective 1 Organic molecules: Larger, more complex molecules whose structure is based on a backbone of C atoms (always contain C as a major part of their structure).

Distinguish between organic and inorganic molecules.

Introduction to Organic Chemistry Chem. Nat. Substances p8 Chemical Formulae and the Structures of Organic Compounds How does the chemist know the structure or composition of a particular chemical substance? Physical properties enable identification, but give no clues as to the composition or structure.

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